

EFFECT ON SLAG HYDRATION OF BLAST-FURNACE SLAG CEMENT IN DIFFERENT CURING CONDITIONS

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Abstract

In Japan, the usage of blast furnace slag in construction will increase due to environmental considerations. However, the effect of curing condition and curing period on the hydration of cement and slag hydration in blended cements is not clear, nor is the relative contribution of cement and slag hydration to strength and durability development in blended cement. In this research, the cement and slag hydration ratio in blast-furnace slag cement under different curing conditions, such as temperature, relative humidity, and curing period, is measured using the selective dissolution method. It was found that hydration stops if relative humidity is less than 80%, and that if 80% relative humidity is maintained hydration can occur even at temperatures as low as 7 degrees Celsius. On the other hand, the speed of hydration is faster at higher temperatures such as 40 degrees Celsius than at normal temperatures such as 20 degrees Celsius, so it can be concluded that the influence of temperature on blast-furnace slag in blast-furnace slag cement is similar to that of temperature on ordinary Portland cement. The hydration of blast-furnace slag is also influenced by mix proportions such as water-cement ratio and Blaine's fineness of cement. In the case of high water-cement ratio, the hydration ratio of slag will increase and reach a high hydration ratio; however, in the case of low water-cement ratio the hydration of slag will stop at a low hydration ratio at an early age. Using these results showing the hydration ratio of slag with different mix proportions and curing conditions, estimation of the hydration of blast-furnace slag cement under different conditions may be performed.

Originality

Until now, measurement of the hydration ratio of blast-furnace slag in blast-furnace slag cement has been performed under sealed condition – which represents the ideal case – in order to clearly show the slag hydration. However, these conditions are not reflective of the actual conditions of concrete structures, so it is important to estimate the hydration ratio of blast-furnace slag under real conditions considering both varying temperatures and relative humidity. It is also necessary to consider the effect of hydration ratio at varying mix proportions and different types of blast furnace slag.

Chief contributions

Concrete structure is exposed to various environment such as drying condition, high relative humidity and different temperatures. And also internal condition in concrete structure is influenced high temperature by the heat of cement hydration. Estimation of the hydration of blast-furnace slag in blast-furnace slag cement on different condition is necessary for estimating the properties and durability of concrete structures. Currently, it is possible to estimate durability properties such as drying and autogeneous shrinkage, strength for concrete structures using ordinary Portland cement from hydration of cement on mix proportions, but it is difficult to apply the same methods when using blast-furnace slag cement, because it is different of cement hydration and slag reaction. This research may contribute to increasing the usage of blast-furnace slag in construction and thus helping to protect the global environment.

Keywords: Blast-furnace slag cement, slag reaction, selective dissolution, curing temperature, curing relative humidity

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1. Introduction

In recent years, the usage of blast-furnace slag cement for concrete structures has increased in order to reduce carbon dioxide emissions and for improving durability. When compared to OPC, such blended cement produces smaller diameter pores due to the cement hydration products, and controlling the cement hydration is very important to ensure the concrete structure's durability and strength. However, the mechanism of hydration or reaction in blast-furnace slag cement is not clear – particularly the reaction of slag in blended cement – so it is necessary to clarify this mechanism. In addition, curing is very important to achieve adequate binder hydration. Construction is generally conducted outdoors, where curing conditions and environmental conditions have a large influence, so it is also necessary to consider the effects of temperature and humidity on the hydration mechanism.

This research investigated the influence of curing temperature, mix proportion, and slag replacement ratio on the slag reaction in blast-furnace slag cement under sealed conditions during Steps 1 and 2. In Step 3 of this research, the effect of curing conditions on cement hydration under various temperatures and humidity conditions was investigated and compared between OPC and blast-furnace slag cement.

2. Experimental Outline

2.1 Material

Ordinary Portland cement (OPC) and Ground Granulated Blast-Furnace slag (GGBFS) were used in this research. Table-1 shows the mineral components of these materials.

Table-1 Chemical components of the experimental materials

	Gravity	Blaine	Chemical Components (%)												
	(g/cm ³)	(cm ² /g)	ig.loss	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	CaO	MgO	TiO ₂	MnO	SO ₃	Na ₂ O	K ₂ O	P ₂ O ₅	Cl
OPC	3.15	3410	2.31	20.73	5.54	2.83	64.28	1.00	0.29	0.10	2.16	0.20	0.40	0.17	0.010
GGBFS	2.89	4760	1.31	32.50	13.72	0.26	42.36	6.00	0.75	0.26	2.08	0.22	0.29	0.03	0.006

2.2 Samples

In Steps 1 and 2 the cement paste sample was made using 30 milliliter glass bins, and sealed curing conditions, such as not demolded, were set for the total experimental age. Specimens were prepared in the laboratory at 20 degree Celsius. After preparation samples were capped immediately and exposed to each water curing environment, in which the water temperature was adjusted.

In Step 3, the sample size was 20×30×10 millimeters in order to maintain uniform humidity in each sample, and the samples were exposed to each relative humidity condition for the total experimental total. Samples were prepared in the laboratory at 20 degree Celsius and demolded after 24 hours, then exposed to various curing conditions. However, samples were dried in the atmosphere for 2 to 3 hours before exposure to each curing condition in order to avoid changing the humidity setting due to water evaporation from the samples. The set curing condition parameters were the temperature and humidity.

2.3 Parameters

Factors affecting the hydration or slag reaction were varied, focusing particularly on material component such as slag replacement at different temperatures, water-binder ratios and curing conditions. Tables 2 to 4 show the parameters of this research. The samples used were cement paste. Steps 1 and 2 targeted the measurement of the hydration ratio of cement and GGBFS under sealed, or ideal, conditions. Step 3 focused on the measurement of the hydration ratio of cement and GGBFS under environmental conditions.

Step 1 (Table-2) focused on the effect of curing temperature and GGBFS replacement on the reaction ratio. The water-binder ratio was constant at 50%, and curing temperatures were 5 degrees Celsius for

winter conditions, 20 degrees Celsius for normal conditions, and 30, 40, and 60 degrees Celsius for summer conditions. The slag replacement ratio was 0% (OPC), 20% (blast-furnace slag cement type BA), 42% (BB) and 67% (BC).

Step 2 (Table 3) focused on the effect of mix proportion and slag replacement ratio on the reaction ratio under normal curing temperatures at 20 degree Celsius. Three water binder ratios were set: 30%, 50%, and 60%; and there were three GGBFS replacement ratios: 0%, 20/30%, 50%, and 70%.

Step 3 (Table 4) focused on the effect of relative humidity and temperature on the reaction ratio. The water binder ratio was 50% and slag replacement was 50%. The curing temperature was 7.5, 20 and 40 degrees Celsius, and four curing relative humidity values were selected: 40%, 60%, 80% and sealed conditions. Humidity was controlled by sodium hydroxide in order to preventing changes to the pore distribution and cement hydration due to carbonation. The humidity control error was $\pm 3\%$.

Table-2 Parameters of Step 1 (curing temperature)

		Replacement (%)			
		N	BA	BB	BC
		0%	20%	42%	67%
Temperature condition (°C)	5	○	○	○	○
	20	○	○	○	○
	30	○		○	
	40	○	○	○	○
	60	○	○	○	○

Table-3 Parameters of Step 2 (mix proportion)

		Replacement				
		N	BA		BB	BC
		0%	20%	30%	50%	70%
W/B	L(30%)	L0	—	L30	L50	L70
	M(50%)	M0	M20	—	M50	M70
	H(60%)	H0	—	H30	H50	H70

Table-4 Parameters of Step 3 (curing conditions)

		Relative Humidity			
		40%	60%	80%	Sealed
Temp.	7.5°C		○	○	○
	20°C	○	○	○	○
	40°C		○	○	○

2.4 Methods for experiment

The hydration ratio and hydration products were measured using the following method.

(1) Sample preparation

The gathered samples were crushed at the testing age and soaked in acetone for 24 hours to stop hydration. For Steps 1 and 2, the testing ages were 1, 3, 7, 28, 56 and 91 days; for Step 3, the testing ages were 1, 3, 7 and 28 days.

(2) Cement hydration

The cement hydration was measured by ignition loss weight at 700 degree Celsius, which includes evaporation of free and chemically-bound water for hydration with cement and GGBFS. Oxidized slag is also protected in high temperature environments, which is why 700 degree Celsius was selected.

(3) Reaction ratio of blast-furnace slag

The reaction ratio was measured by the selective dissolution method using salicylate acid – acetone – methanol solution. It is possible to measure the quantities of an-hydrated blast-furnace slag in this method. The grained sample (0.5g) was mixed with salicylate acid (2.5g), acetone (35ml) and methanol (15ml) for 1 hour, then allowed to sit for 23 hours. The ratio of slag hydration was calculated by using Equation [1] from the result for quantity of an-hydrated slag.

$$\alpha(t) = 100 - \frac{x(t) \cdot (100 - Ig'(t)) - mk_1k_2(100 - Ig(t))}{mk_3k_4(100 - Ig(t))} \times 100 \quad [1]$$

Where : t= hydration age, $\alpha(t)$ = ratio of slag hydration (%),
x(t)= amount of insolutive samples (mg), m=sample (mg),
Ig(t) = ignition loss of hydrated samples (%),
Ig'(t)=ignition loss of insolutive samples(%),
k1=Content ratio of Ordinary Portland Cement for calculated non-hydrated(%),
k2=ratio of insolutive only OPC(%),
k3= Content ratio of Blast-Furnace slag for calculated non-hydrated(%),
K4= ratio of insolutive only Blast-Furnace slag(%)

(4) Quantitative analysis of calcium hydroxide by TG-DTA

Quantitative analysis of calcium hydroxide was conducted using TG-DTA. The heating rate was assumed to be 30 degrees Celsius per minute. Two measurement results were targeted: (1) for obtaining the amount of hydration product, the amount of weight for evaporated water from 20 to 200 degrees Celsius was calculated; (2) for obtaining the cement hydration ratio, the weight loss for evaporated water from 20 to 700 degrees Celsius was calculated. 700 degrees Celsius was assumed due to the influence of the oxidation of S included in GGBFS; (3) for obtaining the amount of $Ca(OH)_2$, the point where inclination of curve was different was calculated.

3. Experimental result

3.1 Effect of curing temperature and slag replacement under sealed conditions (Step 1)

Figure-1 shows the total binder ignition loss under varying temperatures for each slag replacement. At low temperatures (5 degrees Celsius), the hydration progress was slow. The hydration ratio at low temperatures was the same as at 20 degrees Celsius at 58 days of age; however, the hydration continued over a long period. On the other hand, for high temperature conditions the hydration progress occurred very early and the value of the hydration ratio was saturated at an early age, but the hydration ratio was smaller for high slag replacement than for OPC. Figure-2 shows the slag reaction ratio under different temperature conditions for each slag replacement. The slag reaction ratio followed the same behavior as the cement hydration ratio – that is, when the amount of slag increased the ratio of slag reaction tended to become small.

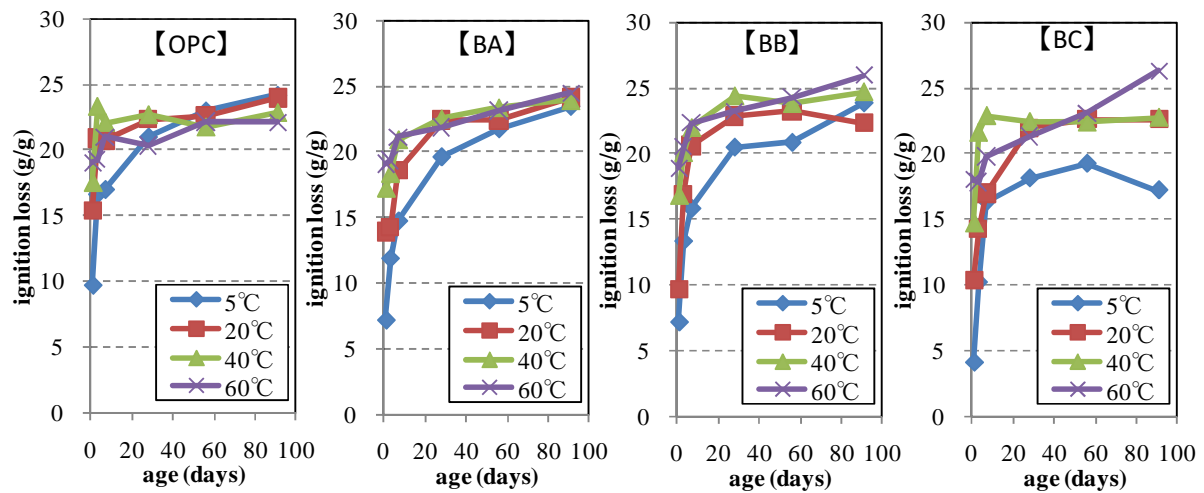


Figure-1 Ignition loss results for varying types of cement and temperatures

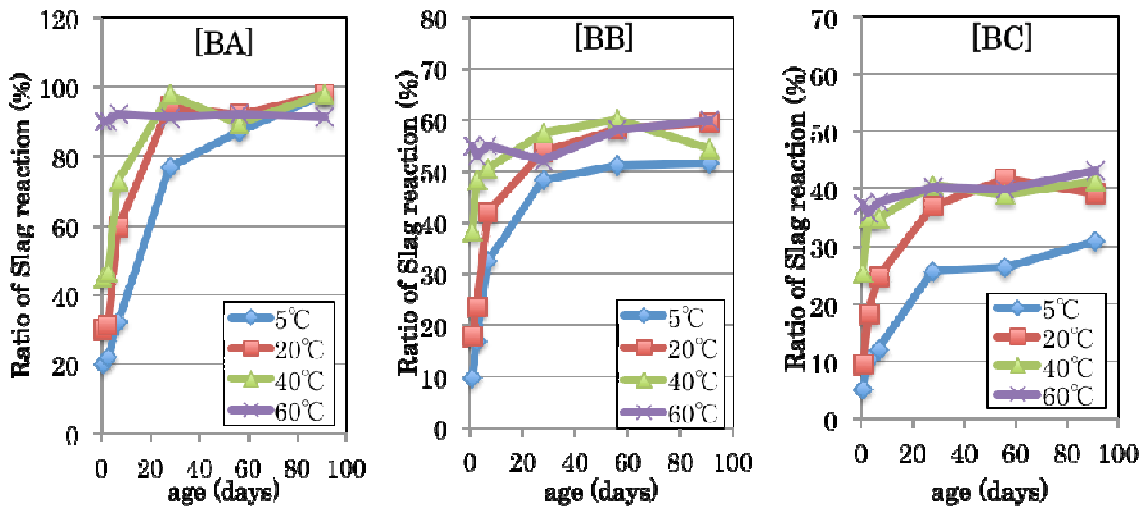


Figure-2 Slag reaction result for varying types of cement and temperatures

3.2 Effect of mix proportions (Step 2)

Figure-3 shows the cement hydration ratio results measured by the ignition loss for each water binder ratio. Total ignition loss of OPC (L0, M0 and H0) – that means the evaporated water from combined water at 700 degree Celsius – was smaller than that of blended cement. For the low water-cement ratio, cement hydration was stopped at early age due to lack of water. On the other hand, in the case of high water-cement ratio cement hydration progressed with curing age.

Figure-4 shows the slag reaction results for different water-binder ratios. For the medium and high water-binder ratios, slag reaction progressed with curing age. However, for low water-cement ratio the slag reaction stopped at an early age and with a small value reaction ratio (15-20%). When comparing M and H, in the case of M (W/C=0.50) the progress of slag reaction occurred earlier than that of H, which suggests that may be the best water-binder ratio considering slag reaction.

3.3 Effect of curing humidity (Step 3)

The ignition loss results are shown in Figure-5. Cement hydration under sealed conditions progressed for each cement type and temperature, and the lower the curing humidity, the lower the initial ignition loss. In addition, the cement hydration stopped due to lack of hydration water in dry conditions, as

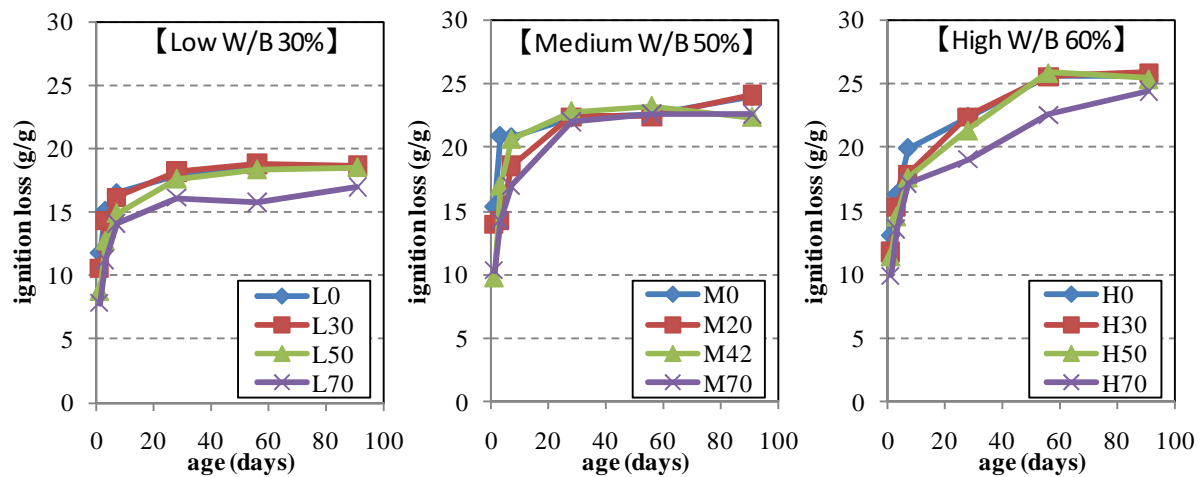


Figure-3 Ignition loss results for varying water-binder ratio and slag replacement

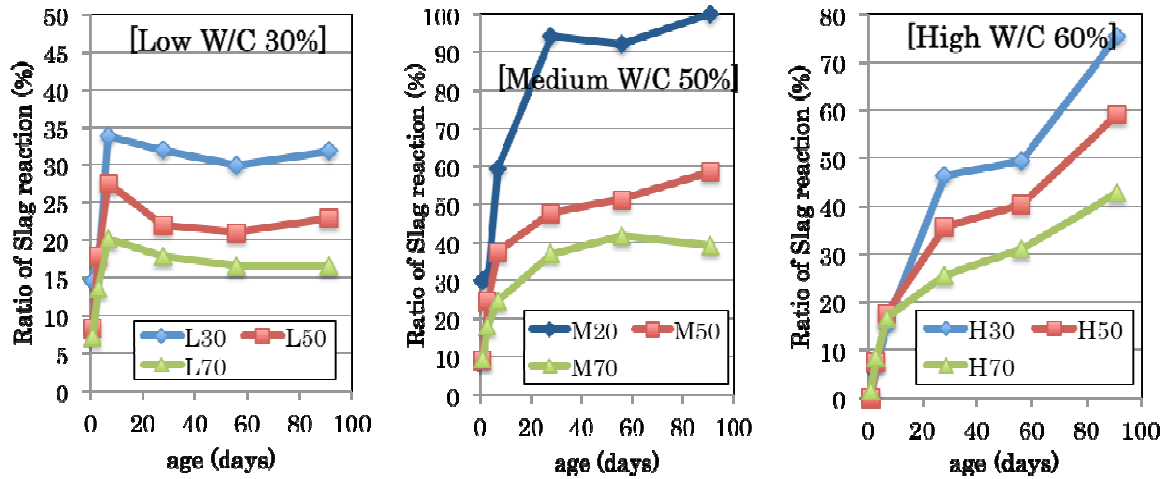


Figure-4 Slag reaction ratio for varying water-binder ratio and slag replacement

shown by the similarly-low hydration ratios of RH40% and RH60%. On the other hand, the hydration of RH80% was lower than the hydration in the sealed case but proceeded over time, so the minimum humidity needed for hydration to proceed was between RH60% and RH80%. Hydration was delayed under low temperatures, but under high temperatures early hydration proceeded but then stopped.

The results of the blast furnace slag reaction by selective dissolution method are shown in Figure-6. Similar to ignition loss, the slag reaction proceeded under high humidity but stopped in dry conditions (RH40% and RH60%). The reaction was also delayed in low temperature and the reaction of RH80% stopped at the level of RH60%. However, at high temperature the results of RH80% were comparable to the results of the sealed conditions.

4. Conclusion

Table-5 shows the summary of the research results.

Table-5 Summary of the research results

		slag reaction			
		← stop	slow	Normal	→ quick
Step 1	Temperature		5°C	20°C	40,60°C
	Slag replacement		BC	BB	BA
Step 2	Water binder ratio	L(30%)		M(50%)	H(60%)
Step 3	Relative humidity	RH40,60%	RH80%	Sealed	

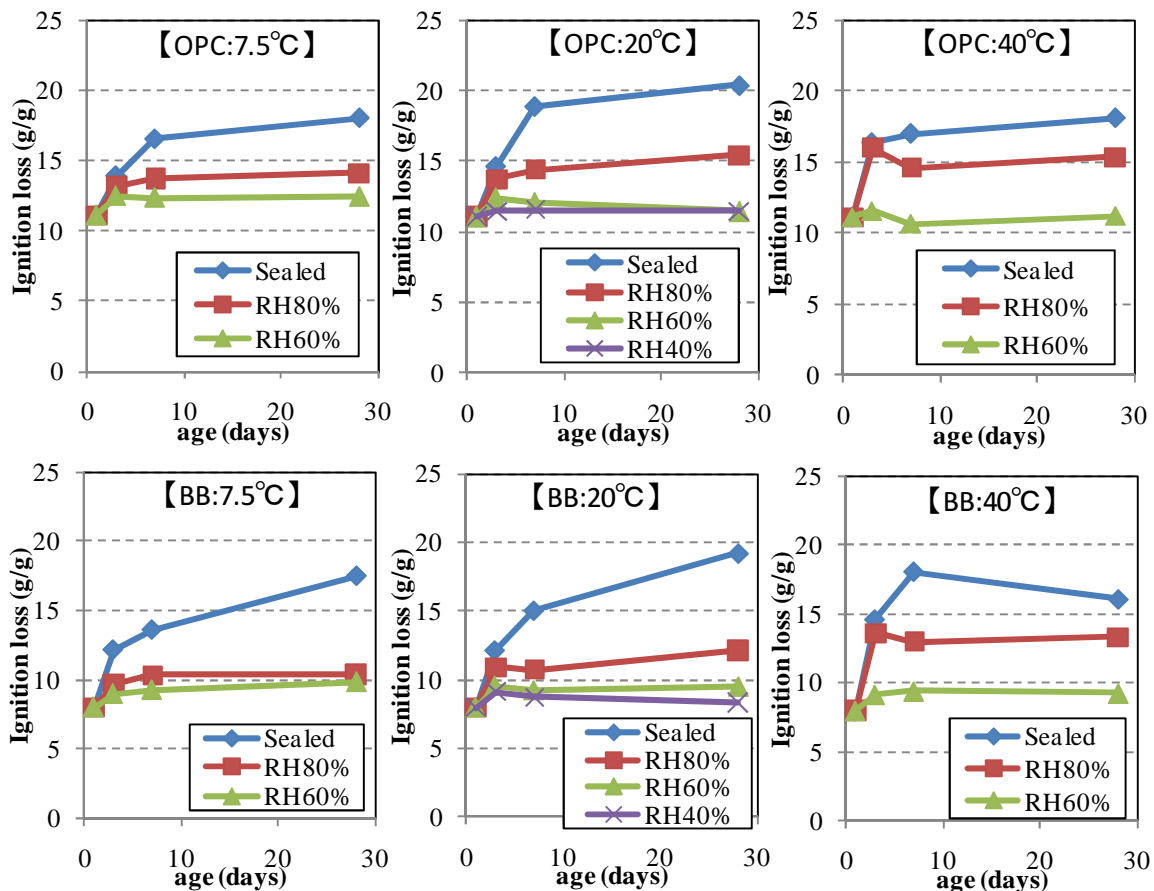


Figure-5 Ignition loss results for varying temperature and relative humidity (Upper: OPC, Lower: Blast-furnace slag cement)

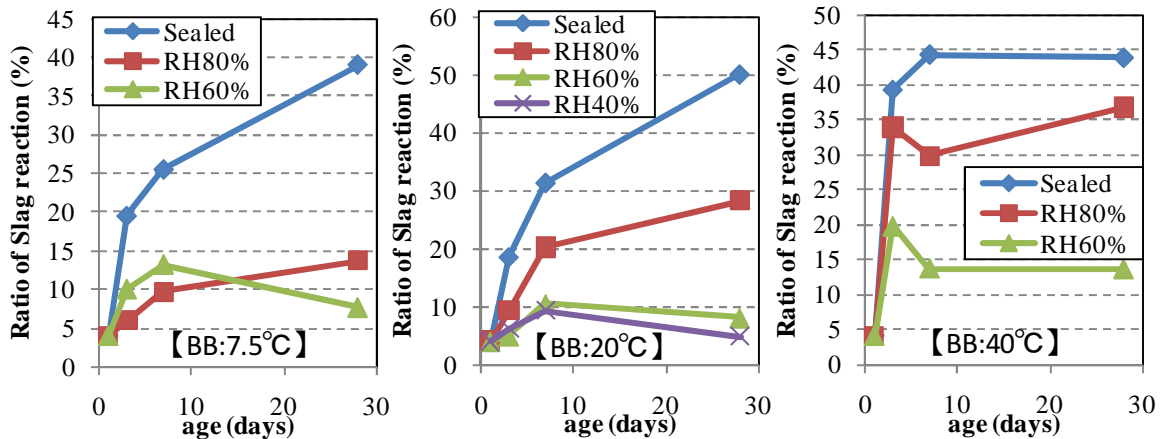


Figure-6 Ratio of slag reaction results for varying temperature and relative humidity

Reference

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